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The contact angle that a liquid–fluid interface makes with a solid surface is a fundamental property in colloid science and interfacial phenomena. This entry works to present some of the most widely used methods to measure macroscopic contact angles.

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Surface area is more important than size for defining colloids. They behave differently than classic objects, or macroscopic bodies, because of their high surface-to-volume ratio, or more accurately, surface-to-mass. The force due to gravity acts on the mass while intermolecular forces effectively act at particle surfaces. If these interparticle or surface forces are large enough to alter the effects of gravity on an object, then it has the characteristics of a colloid.

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