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tends to minimize the surface... any substance  
that interferes with the hydrogen bonding  
betwee... water that contains dissolved  
substances. the dissolving medium in a  
solution. Page 2/5

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SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449) This section describes the



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properties of water in the liquid and solid states and explains how hydrogen bonding affects the surface

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445–449) This section describes the  
properties of water in the liquid and solid  
states and explains how hydrogen bonding  
affects the surface tension and vapor  
pressure of water.

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Key Concepts 15.1 The high surface tension of water and low vapour pressure are due to the hydrogen bonding between the molecules The structure of ice is a regular open frame-work of water molecules held together by hydrogen bonds and arranged like a honeycomb Molecules in liquid water are packed closer together than they are in ice

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