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Chapter 8 - Chemical
Equations and Reactions
8-1 Describing
Chemical Reactions I.
Introduction A.

Reactants 1. Original
substances entering into
a chemical rxn B.

Products 1. The
resulting substances
from a chemical rxn

Reactants \rightleftharpoons Products

C. Chemical Equation 1.
Represents with

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symbols and formulas,
the identities and
relative amounts of the
Reactions
...

Answer Key

~~Chapter 8 – Chemical
Equations and Reactions~~

Describing Chemical
Reactions Key Terms
chemical equation word
equation precipitate
formula equation
coefficient reversible
reaction A chemical

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Chemical reaction is the process by which one or more substances are changed into one or more different substances. In any chemical reaction, the original substances are known as the reactants, and the resulting substances are known as the products.

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Reactions The evolution of energy as light and heat is an indication that a chemical reaction is taking place. CHAPTER 8 Fireworks.

CHEMICAL EQUATIONS AND REACTIONS 261 SECTION 1 O

OBJECTIVES List three observations that suggest that a chemical reaction has taken place.

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Equations and Reactions~~

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Concepts: Terms in this
set (36) Chemical

Equation. an equation

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that represents, with symbols and formulas, the identities and relative molecular or molar amounts of the reactants and products in a ...

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Cumbustion reaction
Practice balancing
equations. Types of

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chemical reactions.

What type of reaction is it? Practice balancing equations. The reactivity series of metals.

Chapter labs: Evidence for chemical change

Chapter homework:

Section 1: Chapter review 1 thru 10.

Section 2: Chapter review 18 thru 21.

Section 3: Chapter review 31 thru 33 ...

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~~Chapter Eight
[Chemical Equations
and Reactions]~~

CHAPTER 8 REVIEW Chemical Equations and Reactions SECTION 2 SHORT ANSWER

Answer the following questions in the space provided. 1. Match the equation type on the left to its representation on the right. c synthesis (a)

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$AX + BY \rightarrow AY + BX$ d

decomposition (b) $A + BX$

$\rightarrow AX + B$ b single-

displacement (c) $A + B \rightarrow$

AX a double-

displacement (d) $AX \rightarrow$

$A + X$

~~8 Chemical Equations and Reactions~~

Chapter 8: Chemical
Equations and
Reactions. I. Describing
Chemical Reactions. A.

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A. chemical reaction. is the process by which one or more substances are changed into one or more different substances. A. chemical. equation.

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Reactions

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Answer Key~~

8-2 Balancing Chemical Equations. Atoms rearrange in chemical reactions, but atoms cannot be created or destroyed (Law of Conservation of Matter). For this reason, the number of each kind of

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atom...

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Equations and Reactions
CP Chemistry~~

druckerja. Holt
Chemistry Chapter 8
Chemical Equations and
Reactions. the process
by which one or more
substa. release of
energy as heat, release
of e. a representation of
a chemical reaction. the

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substance is a solid.

chemical reaction. name
some signs that a
chemical reaction has
taken place. chemical
equation.

~~chapter 8 test chemistry
reactions chemical
equations ...~~

Chapter 8: Chemical
Equations and
Reactions. To balance a
chemical equation, it is

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permissible to adjust
the 1. coefficients. 2.
subscripts. 3. formulas of
the products. 4. number
of products., 2. In a
chemical equation, the
symbol (aq) indicates
that the substance is 1.
water. 2. dissolved in
water. 3. an acid. 4.
insoluble., 3.

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Equations and Reactions~~

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Chapter 8 Chemical

Reactions Worksheet 1.

the equation must

represent known facts 2.

the equation must

contain the correct

formulas for the

reactants and products

3. the law of

conservation of mass

must be satisfied. A

small whole number that

appears in front of a 3.

the law of conservation

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of mass must be
satisfied.

Equations

Reactions

~~Chapter 8 Lesson 2~~

~~Chemical Equations~~

~~Reactions Answers ...~~

Chapter 8 is the first chapter of the second semester. Here, we will cover chemical reactions and equations. Students will first learn about the basics of chemical reactions and

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how to recognize...

~~Chapter 8 Chemical Equations & Reactions~~

~~Answer Key~~

1. To balance a chemical equation, it is permissible to adjust the

- a. coefficients.
- b. subscripts.
- c. formulas of the products.
- d. number of products.

2. In a chemical equation, the symbol (aq)

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indicates that the
substance is a. water. b.
dissolved in water. c. an
acid. d. insoluble., 3.

The production of a
slightly soluble solid
compound in a double-
displacement reaction
results in the ...

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Solved Nom Balancing
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Important Questions For

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Chapter 8 Redox ...

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Test Answer Key ...~~

Name _____ Date

_____ Chapter 11 Test:
Stoichiometry 1. Write a
balanced chemical
equation for a reaction
between zinc and copper
II sulfate. 2. If 5.00
grams of zinc reacts

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with 5.00 grams of
copper II sulfate,
determine the limiting
reactant. 3.

Answer Key

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