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01:31. CHAPTER 3 - Problem 1 What
element does this spectra represent?
(A) Boron (B) Nitrogen (C) Aluminum
...

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chemistry exam

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Step 1: $\text{NH}_3(\text{aq}) + \text{OCl}^-(\text{aq}) \rightarrow \text{NH}_2\text{Cl}(\text{aq}) + \text{OH}^-$
Step 2: $\text{NH}_2\text{Cl}(\text{aq}) + \text{NH}_3(\text{aq}) \rightarrow \text{N}_2\text{H}_5^+(\text{aq}) + \text{Cl}^-$
Step 3: $\text{N}_2\text{H}_5^+(\text{aq}) + \text{OH}^- \rightarrow \text{N}_2\text{H}_4(\text{aq}) + \text{H}_2\text{O}(\text{l})$

(a) (i) What is the equation for the overall reaction? To determine the net equation, all three equations

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must be added together, and species that appear on both sides of the arrow can be eliminated.

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Semester 1 - AP Chemistry

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all free-response questions on this page reflect the current exam, but the question types and the topics are similar, making them a ...

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Unfortunately, for AP Chemistry, most of the available official practice materials are for the old version of the test (pre-2013), but these can still be useful for practice. You should be able to get newer practice tests from your teacher or through review books.

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The AP Chemistry Exam format is:

Section I: 60 Multiple-Choice

Questions (90 minutes, 50% of final

grade) Section II: 7 Free-Response

Questions (105 minutes, 50% of final

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grade) There are 3 long-answer questions and 4 short-answer questions. The free-response skills assessed are: Experimental design; Quantitative/qualitative translation

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Exams

May 16, 2017 / in AP Chemistry / by erinnelson. The multiple-choice section of the AP Chemistry test consists of 60 questions that you will have 90 minutes to answer. That's an average of one and a half minutes per question. Refresh yourself on the

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general AP test strategies and multiple-choice strategies, then apply the concepts to the following five AP Chemistry practice questions.

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Throughout the exam the following symbols have the definitions specified unless otherwise noted. L, mL = liter(s), milliliter(s) mm Hg = millimeters of mercury. g = gram(s) J, kJ = joule(s), kilojoule(s) nm = nanometer(s) V = volt(s) ... 2017 AP ®
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